



ABS Material

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INTRODUCTION

Because of a unique balance of properties, modern ABS copolymers are being used on an ever increasing scale for the manufacture of many industrial and domestic products.

The material is very tough and resilient, has high impact strength, good chemical resistance and is non toxic and taint free. These advantageous properties have attracted engineers in many industries to the use of ABS piping systems rather than traditional materials, which do not have these distinctive benefits.

ABS piping systems are replacing many failed piping systems made from other materials.

The Eurapipe ABS system comprises a range of matched pressure pipes and fittings, joined together by a wide variety of methods including cold solvent cement welding or our rubber ring joint system.

THE MATERIAL

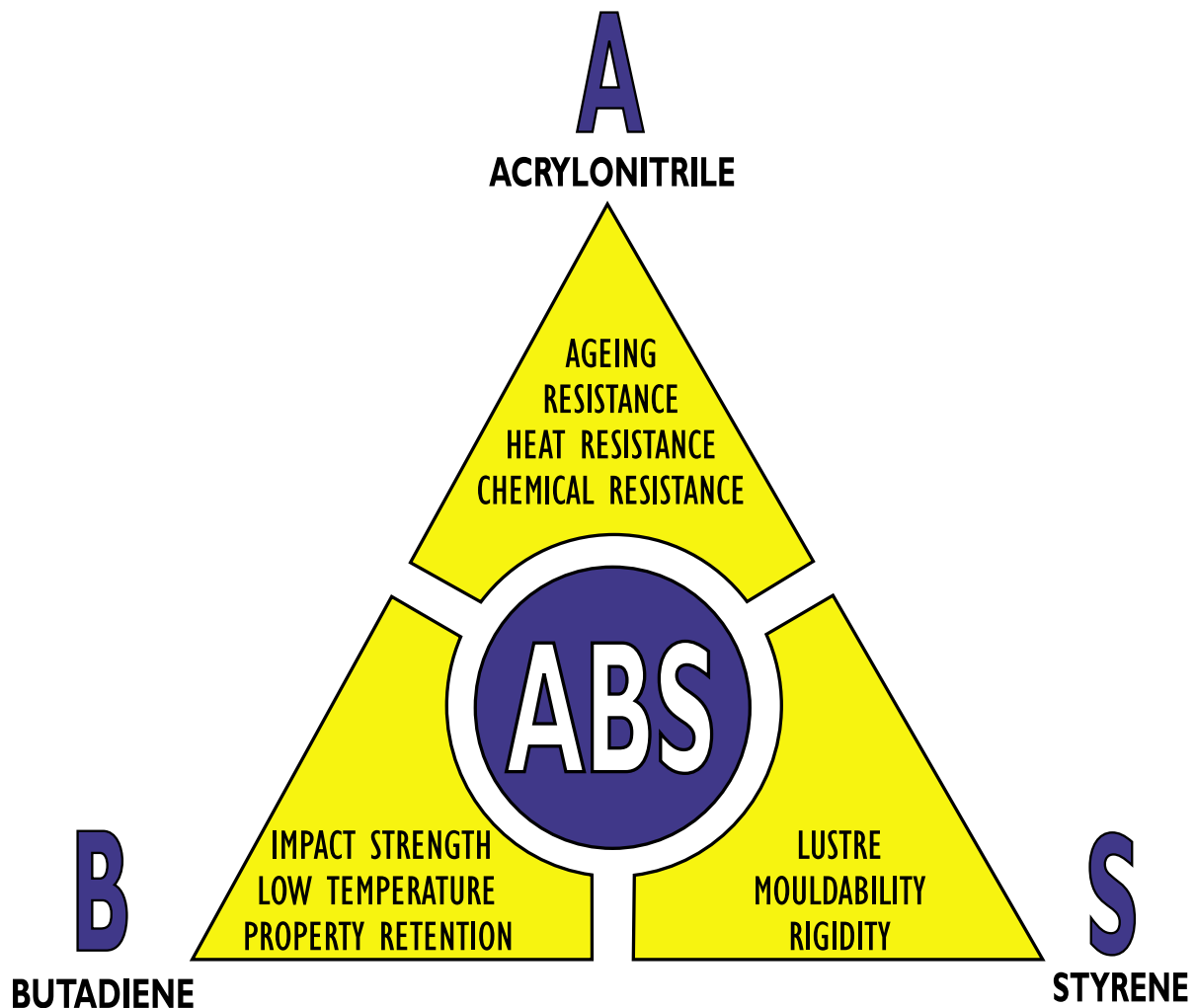
Acrylonitrile - Butadiene - Styrene (ABS) identifies a family of engineering thermoplastics with a broad range of performance characteristics.

The copolymeric system is alloyed to yield the optimum balance of properties suited to the selected end use.

ACRYLONITRILE - imparts chemical resistance and rigidity.

BUTADIENE - endows the product with impact strength, toughness and abrasion resistance.

STYRENE - contributes to the lustre, ease of processing and rigidity.



MATERIALS PROPERTIES

The formulation used by Eurapipe has been developed in conjunction with polymer manufacturers to optimise performance in respect to tensile strength, chemical resistance, ductility, resistance to weathering, heat stability, low toxicity, taint free and ease of processing from raw material to finished product.

ABS is tough and strong over the recommended temperature range of -30°C to +60°C.

The outstanding properties of ABS are:

- ✓ High impact strength and ductility, which combine to give exceptional toughness.
- ✓ Good chemical resistance.
- ✓ Abrasion resistance.
- ✓ High strength solvent weld jointing which allows efficient system assembly and modification.



- ✓ Rubber Ring jointing methods, allowing compatible systems jointing techniques.
- ✓ Nontoxic and non-taint properties.
- ✓ Withstands aggressive ground waters.
- ✓ High strain tolerance for buried applications.
- ✓ Good resistance to ultraviolet light.
- ✓ Lower celerity and extreme tolerance to water hammer surges.

Property*	Reference Temperature	S.I.Unit	Other Units
Ultimate tensile strength (strain rate 50mm/min) ASTM D638 Type I	20 °C	40 MPa	5800 lbf/in ²
Elongation at break	20 °C	50%	50%
Instantaneous Flexural Modulus	20 °C	2200 MPa	319 072 lbf/in ²
Compressive strength	20 °C	42 MPa	6100 lbf/in ²
Izod impact strength (notched) ASTM D256 (method A)	23 °C	340 J/m notch	6.4 ft lb/in notch
Specific gravity		1.05 x 10 ³ Kg/m ³	65.5 x 10 ⁻³ lb/ft ³
Vicat softening point ASTM D1525		95 °C	203 °F
Coefficient of thermal expansion		10.1 x 10 ⁻⁵ m/m°C	5.6 x 10 ⁻⁵ ft/ft°F
Maximum operating temperature		60 °C	140 °F
Poisson's ratio		0.35	
Thermal conductivity		0.2 W/m°C	1.3 BTU/ft ² /in/°F
Specific heat		1.47 KJ/Kg°C	0.35 BTU/lbm/°F
Volume resistivity		3.5 x 10 ¹⁶ Ω cm	
Dielectric constant		3.20 @ 60 Hz 3.12 @ 10 ³ Hz 2.90 @ 10 ⁶ Hz	

*Test pieces machined from moulded specimens yielded to the above mentioned typical properties

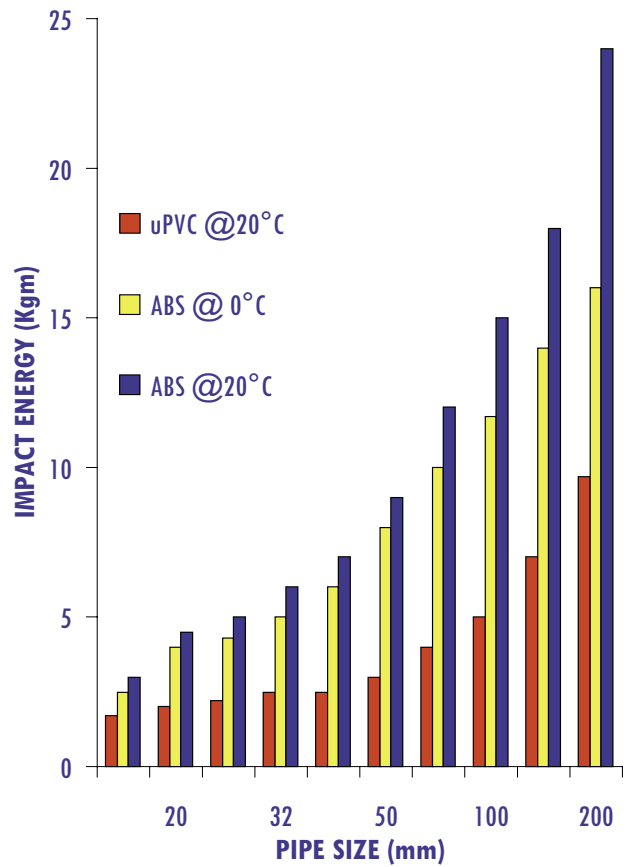
IMPACT STRENGTH

ABS is a relatively ductile thermoplastic, which exhibits very high impact strength compared to other thermoplastics such as uPVC particularly at low temperatures. It is for this reason ABS is used in demanding applications requiring exceptionally high impact strength material such as construction site safety helmets.

As part of the Eurapipe Quality Assurance programme, sample lengths of pipe are routinely impact tested at 0°C as required by AS 3518.

ABS is unique in retaining high levels of impact strength at sub zero temperatures and is significantly superior to most other thermoplastics used in pipe systems.

The graph shows the relatively small reduction in impact strength of ABS between 20°C and 0°C compared with another thermoplastic pipe systems.

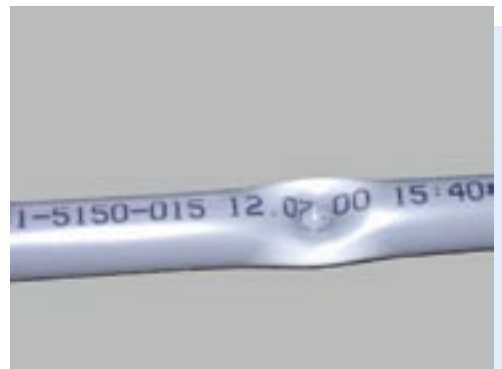


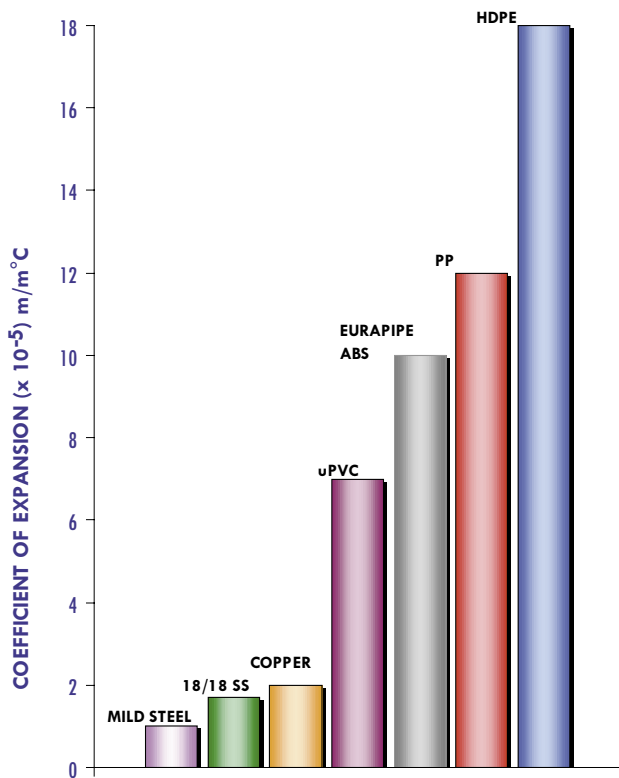
MODE OF FAILURE



ABS is a relatively ductile material and the mode of failure resembles that of soft copper. Failure is by ductile distortion and tearing, the localised nature minimising the loss of pipe contents.

In contrast, crack propagation and hazardous material fragmentation accompany the failure of brittle material.





THERMAL EXPANSION

All thermoplastics expand at a greater rate than metals as shown in the diagram above.

Expansion need not cause undue concern in design or installation of an ABS piping system provided that due recognition is taken at the design stage. The reduced flexural modulus of ABS over that of steel results in reduced loads on supports and equipment arising from thermal strains.

The linear coefficient of thermal expansion of ABS is 10.1×10^{-5} m/m °C.

TOXICITY AND TAINT

ABS is **free** from heavy metal stabilisers such as **lead** which are often used in the processing of other thermoplastic materials. Therefore, there is **no** possibility of any toxic heavy metals substances being leached from the ABS pipe material into the fluid being conveyed by the pipe.

Eurapipe ABS conforms to AS4020 and has been safely used for many years with potable water, grade I distilled water for medical use, renal dialysis fluid and many foods and beverages.

ABS is regarded as taint free and has been used for conveying potable water, beer, soft drinks, caramel, wines, sauces, chocolate, custard cream and other similar products. It is recommended that food and drink manufacturers test for taste tainting on their own product before installation commences.

RIGIDITY AND STIFFNESS

ABS is classified as a rigid thermoplastic over its working temperature range -30°C to $+60^{\circ}\text{C}$.

With increased temperature, pipe rigidity decreases thus necessitating more frequent support.



WEATHERING

Eurapipe ABS piping systems are suitable for external installation under extreme conditions without additional surface protection.

When ABS products are exposed to the weather, they will suffer some minor degradation of the exposed surface. The degradation results in a reduction of surface gloss, and shift in surface colour to light grey. The degradation is confined to the exposed surface only.

The effect of long-term exposure to sunlight over prolonged periods has minimal effect on the physical properties of ABS systems.

Because of the relatively high flexural modulus of ABS, the stresses induced in a component whilst in service result in smaller strains, therefore minimising the possibility of environmental stress cracking of the exposed surface.



This resistance to failure is further improved by the inherently high impact strength of ABS, particularly at low temperatures, and the ability of the polymer to withstand long term heat exposure with little change to physical properties.



ABRASION RESISTANCE

ABS piping systems have long been successfully employed in applications where abrasion resistance is the prime consideration. The conveying of slurries in the mining, food, power generation and waste water industries is a typical example where ABS has been demonstrated to outlast steel and stainless steel pipes previously employed.

The chemical resistance of ABS combined with impact resistance makes it an ideal choice for such corrosive and erosive environments.

It is these conditions which lead to reduced life of metal pipe systems.

The rubber-like butadiene phase in ABS provides this piping material with outstanding resistance to abrasive media.

Eurapipe sales engineers have the experience to advise on the suitability of ABS pipe for slurry or abrasive applications.

For gravity flow systems the long term low surface roughness enables less steep slopes to be used. Lower slopes can mean reduced building heights which has a great effect on capital costs. Additionally, lower slopes reduce transport velocity, which in turn reduces the wearing of the piping material.

CHEMICAL RESISTANCE

The information given on the following pages is based on the recommendations of the manufacturers of the polymers, field experience and subsequent tests by Eurapipe.

The chemical resistance information has been obtained from numerous sources and it is primarily based on plastic material test specimens that have been immersed in the chemical (not combination of chemicals) and on field experience. Under no circumstances is to be assumed that a mixture of individually acceptable chemicals may be safely used with ABS or any other product.



The effect of the combination of chemicals on the ABS components has to be assessed in conjunction with other factors that have a significant impact upon the lifecycle of the system i.e. temperature, internal pressure, flexural stresses, cyclic loads etc. Any chemical attack is increased when temperature or stress are increased or when temperature or stress are varied.

It is the design engineers responsibility to assess the materials and the exposure under such conditions.

Specific data on industrial chemical applications of ABS can be given by the Eurapipe organisation. Such enquiries are invited for applications not shown here.

Under no circumstances is it to be assumed that a mixture of individually acceptable chemicals may be safely used with ABS or any other product.

Absence of notation indicates the substance has not been tested.

QUICK REFERENCE CHEMICAL RESISTANCE	
Chemical	Resistance
Weak acids	Good resistance
Strong acids	Limited resistance
Weak alkalis	Good resistance
Strong alkalis	Good resistance
Aggressive soils	Excellent resistance
Metal salts	Good resistance
Sea water	Excellent resistance
Aromatic hydrocarbons	Poor resistance
Organic solvents	Poor resistance

Unless stated, all concentrations are 100% or saturated aqueous solution. Reference to saturated solutions is at 20°C.

Resistance Key Information

1. RESISTANT=Little or no attack
2. CONDITIONAL RESISTANCE=Some attack, however may still be suitable when used with a higher pipe class or reduced service life.
3. NOT RECOMMENDED=Little or no resistance. Not suitable for use with ABS pipe.
4. REFER TO EURAPIPE

The information given here is based upon various sources available at the time this manual was created. We reserve the right to revise this information from time to time in the light of subsequent research and experience. The information is to be used as a general guide and there is no warranty or representation, either expressed or implied, that this data is free from errors.

We shall not be liable for any damages of any kind that may result from the use of this data.

Chemical or Agent	Formula	Concentration (%W/V)	Working temperature	
			20°C	50°C
Acetamide	CH ₃ CONH ₂	%	1	
Acetic Acid	CH ₃ COOH	Up to 10	1	1
		10-20	2	
		Over 20 (including Glacial)	3	3
Acetone	CH ₃ COCH ₃		3	3
Acetyl Chloride	CH ₃ COCl		3	3
Alcohols:				
Allyl	CH ₂ =CHCH ₂ OH		3	3
Amyl	CH ₃ (CH ₂) ₃ CH ₂ OH		3	3
Benzyl	C ₆ H ₅ CH ₂ OH		3	3
Butyl (Butanol)	CH ₃ (CH ₂) ₂ CH ₂ OH		3	3
Ethyl (Ethanol)	CH ₃ CH ₂ OH	Up to 50% aq. soln.	1	1
Ethyl (Ethanol)	CH ₃ CH ₂ OH	95% aq. soln.	3	3
Furfuryl	C ₄ H ₃ OCH ₂ OH		3	3
Methyl (Methanol)	CH ₃ OH		3	3
Iso Propyl (propanol)	(CH ₃) ₂ CHOH		3	3
Alum	Al ₂ (SO ₄) ₃ ·K ₂ SO ₄ ·H ₂ O		1	1
Aluminium Chloride	AlCl ₃		1	1
Aluminum Sulphate	Al ₂ (SO ₄) ₃		1	1
Ammonia Solution	NH ₄ OH	35%	1	1
Ammonium Carbonate	(NH ₄) ₂ CO ₃		1	1
Ammonium Molybdate	(NH ₄) ₆ Mo ₇ O ₂₄ ·H ₂ O		1	1
Ammonium Nitrate	NH ₄ NO ₃		1	1
Ammonium Sulphate	(NH ₄) ₂ SO ₄		1	1
Ammonium Thiocyanate	NH ₄ SCN		1	1
Amyl Acetate	CH ₃ COO(CH ₂) ₄ CH ₃		3	3
Aniline	C ₆ H ₅ NH ₂		3	3
Aromatic Hydrocarbons			3	3
Barium Bromide	BaBr ₂		1	1
Barium Carbonate	BaCO ₃		1	1

1=RESISTANT 2=CONDITIONAL RESISTANCE 3=NOT RECOMMENDED 4=REFER TO EURAPIPE

Chemical or Agent	Formula	Concentration (%W/V)	Working temperature	
			20°C	50°C
Barium Chloride	BaCl ₂		1	1
Barium Hydroxide	Ba(OH) ₂		1	1
Battery Acid	H ₂ SO ₄		1	1
Benzene	C ₆ H ₆		3	3
Benzoic Acid	B ₆ H ₅ COOH		3	3
Boric Acid	H ₃ BO ₃		1	1
Brake Fluids			3	3
Brine	NaCl _H 2O	Saturated	1	1
Bromic Acid	HbrO ₃		1	1
Bromine (Gas + Liquid)	Br ₂		3	3
Butane Gas	C ₄ H ₁₀		1	1
Butyric Acid	C ₃ H ₇ COOH	20% aqueous	3	3
Calcium Compounds	Refer to respective sodium compound			
Carbon Dioxide	CO ₂	40% aq. soln.	1	1
Carbon Disulphide	CS ₂	95% sq. soln.	3	3
Carbon Monoxide	CO		1	1
Carbon Tetrachloride	CCl ₄		3	3
Castor Oil			1	1
Chlorine Gas Dry	Cl ₂		2	3
Chlorine Wet			3	3
Chlorine Aqueous Solution		Up to 3% free chlorine	1	1
		Over 3% free chlorine	4	4
Chlorobenzene	C ₆ H ₅ Cl		3	3
Chloroform	CHCl ₃		3	3
Chromic Acid	CrO ₃ +H ₂ O	10%	2	3
		25%	3	3
Citric Acid	HOC(COOH)(CH ₂ COOH) ₂ H ₂ O		1	1
Cresols	C ₆ H ₄ (OH)CH ₃		3	3
Copper Chloride	CuCl ₂		1	1
Copper Fluoride	CuF ₂		1	1
Copper Sulphate	CuSO ₄		1	1

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Chemical or Agent	Formula	Concentration (%W/V)	Working temperature	
			20°C	50°C
Creosote			3	3
Cyclohexane	C ₆ H ₁₂		3	3
Detergents			4	4
Dextrose	C ₆ H ₁₂ O ₆ H ₁₂		1	1
Dichloroethane	CH ₂ ClCH ₂ Cl		3	3
Dichloromethane	CHCl ₂		3	3
Diethylamine	(C ₂ H ₅) ₂ NH		3	3
Diethyl Ether	C ₂ H ₅ OC ₂ H ₅		3	3
Ethylene Glycol	HOCH ₂ CH ₂ OH		1	1
Ferric Chloride	FeCl ₃		3	3
Ferric Nitrate	Fe(NO ₃) ₃		1	1
Ferrous Chloride	FeCl ₂	Saturated	1	2
Ferrous Sulphate	FeSO ₄	40% aqueous	1	1
Formaldehyde (Formalin)	HCHO (+H ₂ O)	10%	1	1
Formic Acid	HCOOH	3%	1	3
Freon	R11, R12, R22, R113, R114		4	4
Fruit Juices			1	2
Gelatine			1	1
Glucose	C ₆ H ₁₂ O ₆		1	1
Glycerine	HOCH ₂ -CHOH-CH ₂ OH		1	1
Hydrochloric Acid	HCl	0-10%	1	1
		10-30%	1	1
		30%-37%	1	3
		>37%	3	3
Hydrofluoric Acid	HF	0-10%	1	2
		>10%	3	3
Hydrofluorosilicic Acid	H ₂ SiF ₆		3	3
Hydrogen	H ₂		1	3
Hydrogen Peroxide	H ₂ O ₂	1%	1	1
		3%	1	2
		5%	1	3
		10% (30 vol)	3	3

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Chemical or Agent	Formula	Concentration (%W/V)	Working temperature	
			20°C	50°C
Iodine Solution in KI	I ₂		1	3
Kerosene			3	3
Ketones			3	3
Lanolin			1	1
Lead Acetate	Pb(CH ₃ OO) ₂		1	1
Linseed Oil			1	3
Magnesium Compounds	Refer to respective sodium compound			
Mesityl Oxide	(CH ₃) ₂ C=CHCOCH ₃		3	3
Methane	CH ₄		1	3
Methoxyethanol	CH ₃ OCH ₂ CH ₂ OH		3	3
Methyl Acetate	CH ₃ COOCH ₃		3	3
Methyl Cyclohexanone	C ₈ H ₉ CH ₃ O		3	3
Methyl Ethyl Ketone	CH ₃ COCH ₂ CH ₃		3	3
Methyl Methacrylate	CH ₂ C(CH ₃)COOCH ₃		3	3
Methylated Spirits			3	3
Milk			1	1
Mixed Acids Limited resistance Dependent on Concentrations			4	4
Molasses		Commercial	1	1
Nickel Sulphate	NiSO ₄	1%	1	1
Nitric Acid	HNO ₃	1%	1	3
		5%	2	3
Nitrogen	N ₂	3	1	1
Oleic Acid	C ₈ H ₁₇ -CO=CH-		1	3
Oxalic Acid	HO ₂ CCO ₂ H		1	4
Oxygen	O ₂		1	1
Ozone	O ₃	20PPM Solution	1	1
		Saturated Solution	3	3
		Gaseous	3	3

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Chemical or Agent	Formula	Concentration (%W/V)	Working temperature	
			20°C	50°C
Petrol			3	3
Phenol	C ₆ H ₅ OH		3	3
Potassium Compounds Refer to respective Sodium compounds				
Propane	C ₃ H ₈		1	1
Pyridine	C ₅ H ₅ N	Trace	3	3
Soap solutions (aqueous)			1	1
Sodium Acetates	Na(CH ₃ COO)		1	1
Sodium Borate	Na ₂ B ₄ O ₇		1	1
Sodium Carbonate	NaCO ₃		1	1
Sodium Chlorate	NaClO ₃		1	1
Sodium Chloride	NaCl		1	1
Sodium Chromate	Na ₂ CrO ₄		1	1
Sodium Cyanide	NaCN		1	1
Sodium Ferrocyanide	Na ₄ F ₆ (CN) ₆		1	1
Sodium Fluoride	NaF		1	1
Sodium Hydrogen Carbonate	NaHCO ₃		1	1
Sodium Hydrogen Sulphate	NaHSO ₄		1	1
Sodium Hydrogen Sulphite	NaHSO ₃		1	1
Sodium Hydroxide	NaOH	Saturated	1	1
Sodium Hypochlorite	NaOCl	>3% available chlorine	3	3
Sodium Iodide	NaI		1	1
Sodium Nitrate	NaNO ₃		1	1
Sodium Permanganate	NaMnO ₄		3	3
Sodium peroxide	Na ₂ O ₂		3	3
Sodium Persulphate	Na ₂ S ₂ O ₈		1	1
Sodium Phosphate	Na ₄ P ₂ O ₇		1	1
Sodium Salicylate	NaC ₇ H ₅ O ₃		1	1
Sodium Silicate	NaSiO _{1.33} H ₂ O		1	1

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Chemical or Agent	Formula	Concentration (%W/V)	Working temperature	
			20°C	50°C
Sodium Sulphate	Na ₂ SO ₄		1	1
Sodium Sulphite	Na ₂ SO ₃		1	1
Sodium Sulphide	Na ₂ S		1	1
Sodium Thiosulphate	NaS ₂ O ₄		1	1
Stannic Chloride	SnCl ₄		1	3
Stannous Chloride	SnCl ₂		1	3
Sulphur Dioxide (Gas)	SO ₂		1	2
Dry			1	2
Wet			1	2
Sulphuric Acid	H ₂ SO ₄	Under 30%	1	1
		30%-50%	1	2
		50%+	3	3
Toluene	C ₆ H ₃₋₅ CH ₃		3	3
Trichlorobenzene	C ₆ H ₃ Cl ₃		3	3
Trichloroethylene	Cl ₂ C=CHCl ₃		3	3
Triethanolamine	N(CH ₂ CH ₂ OH) ₃		1	3
Triethylene Glycol (Trigol)	HOCH ₂ O) ₂ CH ₂ CH ₂ OH		1	2
Turpentine			3	3
Uric Acid	CO(NH) ₂ COC ₂ CO(NH) ₂		1	2
Urine			1	1
Vegetable Oils			1	2
Vinegar			1	2
<u>Water</u>	H ₂ O		1	1
Chlorinated			1	1
Deionized			1	1
Distilled			1	1
Fresh			1	1
Sea			1	1
Wines			1	2
Xylene	C ₆ H ₄ (CH ₃) ₄		3	3
Zinc Orthophosphate	Zn ₃ (PO ₄) ₂		2	2
Zinc Stearate	Zn(C ₁₈ H ₃₅ O ₂) ₂		1	1

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